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solutes and solvents. In this chapter,

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we will focus on solution where the solvent is water. An aqueous solution is water that contains one or more dissolved substance. The dissolved substances in an aqueous solution

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that contains one or more dissolved substance. The dissolved substances in an aqueous solution may be solids, gases, or other liquids.

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base, accepting  $H^+$  from  $HCl$  due to attraction between  $H^+$  and lone pair of  $e^-$  in  $O$

$$HCl(aq) + H_2O(l) \rightarrow Cl^-(aq) + H_3O^+(aq)$$

Ex. 2: Water acts as acid, donating  $H^+$  to  $NH_3$

$$NH_3(aq) + H_2O(l) \rightarrow NH_4^+(aq) + OH^-(aq)$$

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water as the solvent. A solution is a



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